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                MATHDI removed from STN
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                CA/CAplus-Canadian Intellectual Property Office (CIPO) added
                 to core patent offices
        OCT 13
                New CAS Information Use Policies Effective October 17, 2005
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        OCT 17
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        NOV 30
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        DEC 05
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        DEC 14 2006 MeSH terms loaded in MEDLINE/LMEDLINE
NEWS 15
        DEC 14 2006 MeSH terms loaded for MEDLINE file segment of TOXCENTER
NEWS 16
        DEC 14 CA/CAplus to be enhanced with updated IPC codes
NEWS 17
        DEC 16 MARPATprev will be removed from STN on December 31, 2005
NEWS 18
        DEC 21 IPC search and display fields enhanced in CA/CAplus with the
                IPC reform
         DEC 23 New IPC8 SEARCH, DISPLAY, and SELECT fields in USPATFULL/USPAT2
NEWS 19
             DECEMBER 02 CURRENT VERSION FOR WINDOWS IS V8.01,
NEWS EXPRESS
              CURRENT MACINTOSH VERSION IS V6.0c(ENG) AND V6.0jc(jp),
              AND CURRENT DISCOVER FILE IS DATED 02 DECEMBER 2005.
              V8.0 USERS CAN OBTAIN THE UPGRADE TO V8.01 AT
              http://download.cas.org/express/v8.0-Discover/
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=> file reg COST IN U.S. DOLLARS

SINCE FILE TOTAL ENTRY SESSION 0.21 0.21

FULL ESTIMATED COST

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STRUCTURE FILE UPDATES: 26 DEC 2005 HIGHEST RN 870675-00-6 DICTIONARY FILE UPDATES: 26 DEC 2005 HIGHEST RN 870675-00-6

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TSCA INFORMATION NOW CURRENT THROUGH JULY 14, 2005

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Structure search iteration limits have been increased. See HELP SLIMITS for details.

REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

http://www.cas.org/ONLINE/UG/regprops.html

Uploading C:\Program Files\Stnexp\Queries\10716430\form 1 B.str



chain nodes :

1 2 3 4 5 6 7

chain bonds :

1-7 2-5 2-3 2-7 3-4 3-6

exact/norm bonds :

2-5 3-4 3-6

exact bonds :

1-7 2-3 2-7

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS

L1 STRUCTURE UPLOADED

=> d

L1 HAS NO ANSWERS

L1 STR

Structure attributes must be viewed using STN Express query preparation.

=> s 11

SAMPLE SEARCH INITIATED 12:29:15 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 4604 TO ITERATE

43.4% PROCESSED 2000 ITERATIONS

23 ANSWERS

INCOMPLETE SEARCH (SYSTEM LIMIT EXCEEDED)

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**

BATCH **COMPLETE**

PROJECTED ITERATIONS:

88012 TO 96148

PROJECTED ANSWERS:

622 TO 1494

L2 23 SEA SSS SAM L1

=> s 11 full

FULL SEARCH INITIATED 12:29:19 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 91215 TO ITERATE

100.0% PROCESSED 91215 ITERATIONS

1245 ANSWERS

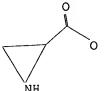
SEARCH TIME: 00.00.01

L3 1245 SEA SSS FUL L1

12/27/2005

=>

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3 1 . 6

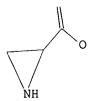
chain nodes :
4 5 6
ring nodes :
1 2 3
chain bonds :
1-4 4-5 4-6
ring bonds :
1-2 1-3 2-3
exact/norm bonds :
1-2 1-3 2-3 4-5 4-6
exact bonds :
1-4

Match level :

1:Atom 2:Atom 3:Atom 4:CLASS 5:CLASS 6:CLASS

L4 STRUCTURE UPLOADED

=> d L4 HAS NO ANSWERS L4 STR



Structure attributes must be viewed using STN Express query preparation.

=> s 14

SAMPLE SEARCH INITIATED 12:29:41 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 11259 TO ITERATE

17.8% PROCESSED 2000 ITERATIONS INCOMPLETE SEARCH (SYSTEM LIMIT EXCEEDED) SEARCH TIME: 00.00.01

4 ANSWERS

FULL FILE PROJECTIONS: ONLINE **COMPLETE**
PROJECTED ITERATIONS: 218822 TO 231538

PROJECTED ITERATIONS: 218822 TO 231538 PROJECTED ANSWERS: 166 TO 734

L5 4 SEA SSS SAM L4

=> s 14 full FULL SEARCH INITIATED 12:29:45 FILE 'REGISTRY' FULL SCREEN SEARCH COMPLETED - 222523 TO ITERATE

100.0% PROCESSED 222523 ITERATIONS 658 ANSWERS

SEARCH TIME: 00.00.01

L6 658 SEA SSS FUL L4

=> file hcaplus

COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION

FULL ESTIMATED COST 322.66 322.87

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=> s 13

L7 2444 L3

=> s 16

L8 364 L6

=> s 13 and 16

2444 L3

364 L6

L9 24 L3 AND L6

=> s base

650906 BASE

148857 BASES

L10 741738 BASE

(BASE OR BASES)

=> s 19 and 110

L11 2 L9 AND L10

=> d ibib abs 1-2

L11 ANSWER 1 OF 2 HCAPLUS COPYRIGHT 2005 ACS on STN
ACCESSION NUMBER:
DOCUMENT NUMBER:
11996:167574 HCAPLUS
124:232231
Aziridine compounds, methods of preparation, and reactions thereof, as intermediates for thiamphenicol and analogs
INVENTOR(S):
DAVIS, Franklin A.: Zhou, Ping: Reddy, Gaddampally Venkat
Venkat
Drexel University, USA
PCT Int. Appl., 62 pp.
CODEN: PIXXD2
DOCUMENT TYPE:
LANGUAGE:
DOCUMENT TYPE:
LANGUAGE:
English
FAMILY ACC. NUM. COUNT:
FAMILY ACC. NUM. COUNT:

PATENT INFORMATION	:			
PATENT NO.	KIN	D DATE	APPLICATION NO.	DATE

WO 9530672	A1	19951116	WO 1995-US4911	19950501
W: AM, A	T. AU. BB.	BG. BR. BY.	CA, CH, CN, CZ, DE	, DK, EE, ES, FI,
			KP, KR, KZ, LK, LR	
MG, N	N, MW, MX,	NO, NZ, PL,	PT, RO, RU, SD, SE	, SG, SI, SK, TJ,
TT, U	A .			
RW: KE, N	W, SD, SZ,	UG, AT, BE,	CH, DE, DK, ES, FR	, GB, GR, IE, IT,
LU, N	C, NL, PT,	SE, BF, BJ,	CF, CG, CI, CM, GA	, GN, ML, MR, NE,
SN, T	D, TG			
US 5789599	A	19980804	US 1994-239097	19940506
AU 9524260	A1	19951129	AU 1995-24260	19950501
PRIORITY APPLN. IN	FO.:		US 1994-239097	A 19940506

OTHER SOURCE(S):

CASREACT 124:232231; MARPAT 124:232231

WO 1995-US4911

19950501

Novel N-sulfinyl-2-carboxy- and N-hydrogen-2-(hydroxymethyl)aziridine compds. I and II and their stereoisomers are provided (wherein R1-R5 = H, hydrocarbyl radicals containing 1-40 C atoms, 0-40 halo atoms, and 0-10 heteroatoms (B, N, O, S, P, Si, Se); both R3 and R4 * H; sulfinyl molety may be racemic or optically enriched]. The asym. synthesis of N-sulfinylaziridines is readily accomplished in high diastereomeric

L11 ANSWER 2 OF 2 HCAPLUS COPYRIGHT 2005 ACS on STN ACCESSION NUMBER: 1994:580151 HCAPLUS DOCUMENT NUMBER: 121:180151 The synthesis and stability of an The synthesis and stability of aziridino-glutamate,

AUTHOR (S):

irreversible inhibitor of glutamate racemase Tanner, Martin E.; Miao, Shichang Dep. Chem., Univ. British Columbia, Vancouver, BC, CORPORATE SOURCE:

121, Can. Tetrahedron Letters (1994), 35(24), 4073-6 CODEN: TELEAY; ISSN: 0040-4039 Journal SOURCE:

DOCUMENT TYPE:

English CASREACT 121:180151

OTHER SOURCE(S):

Aziridino-glutamate (±)-I was synthesized by heating α-fluoromethylglutamate II in base. In neutral solution, (±)-I was shown to cyclize to the y-lactone III with a half life of 4 min. Aziridino-glutamate was shown to irreversibly inactive glutamate racemase by alkylating an active site cysteine residue. Electrospray mass spectrometry was used to establish that a covalent bond had formed and that this bond protects one of the enzyme's two cysteine residues from reacting with iodoacetate under denaturing conditions.

Lll ANSWER 1 OF 2 HCAPLUS COPYRIGHT 2005 ACS on STN (Continued) and good yield by a Darzens-type reaction of a metal enolate of an anhalo ester with an enantiopure sulfinimine. Ring-opening of the aziridines affords a-mino acids and otherwise difficult to prep. syn-B-hydroxy-a-mino acids, both key structural units found in many bioactive materials. The N-sulfinyl radical may be selectively removed from the novel aziridine compds. by treatment with acid or base. Alternatively, the N-sulfinyl radical may be oxidized to provide the corresponding N-sulfonyl-aziridine, or reduced to form the corresponding lH-2-(hydroxymethyl)aziridine, either of which may subsequently be ring-opened to provide precursors to bloactive compds. For example, BrCH2CO2Me was lithiated with (Me3Si)2NLi in THF, and reacted

ted with $(S)-\{+\}-N-benzylidene-p-toluenesulfonimine to give 65% <math>\{2S,3S\}-I$ [R1 = Me, R2 = R4 = H, R3 = Ph, R5 = p-MecG6H4] $\{III\}$, plus 6% $\{2S,3R\}-isomer$ byproduct. The analog of III with R3 = p- $\{MeS\}$ C6H4 was similarly prept, then reduced to the corresponding hydroxymethyl compd. II, hydrolyzed to an aminopropanediol, N-dichloroacetylated, and oxidized with m-clC6H4C(O)OOH, to give the antibiotic thismphenicol.

=> s amine

261513 AMINE

245844 AMINES

L12 399888 AMINE

(AMINE OR AMINES)

=> s 19 and 112

L13

3 L9 AND L12

=> d ibib abs 1-3

Processes for preparing optically active amino acid

derivatives
Sugawara, Masanobu; Fujii, Akio; Okuro, Kazumi; Saka,
Yasuhiro; Nagashima, Nobuo; Inoue, Kenji; Takeda,
Toshihiro; Kinoshita, Koichi; Moroshima, Tadeshi;
Fuse, Yoshihide: Ueda, Yasuyoshi
Kaneka Corporation, Japan; et al.
PCT Int. Appl., 74 pp.
CODEN: PIXXD2
Patent INVENTOR (S):

PATENT ASSIGNEE (S): SOURCE:

DOCUMENT TYPE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

WO 2001060795 A1 20010823 WO 2001-JP1132 20010210
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CR, CU, CZ, DE, DK, DM, CD, EE, ES, FI, GB, GD, CE, GH, CM, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT LU, LV, MA, MD, MG, MC, MN, MM, MC, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VM, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, CB, CG, CI, CM, GA, CM, CW, ML, HR, NZ, SN, TD, TG
CA 2369678 AA 20010823 CA 2001-2369678 20010216
AU 2001023227 A5 20010827 AU 2001-23227 20010216
BY: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, LE, SI, LT, LV, FI, RO
NO 2001005042 A 200110824 NO 2001-5042 20011016
US 2005143586 A1 20050630 CA 2002-926346 PRIORITY APPLN. INFO.: JP 2000-334391 WO 2001-JP1132

OTHER SOURCE(S):

CASREACT 135:195786; MARPAT 135:195786

AB An optically active amino acid derivative is prepared either by subjecting an optically active 3-haloalanine derivative XCH2C*H(NH2)CO2R1 [X is

halogen: Rl
is hydrogen or the like; the asterisk represents an asym. carbon atom] to
N-protection followed by cyclization or cyclization followed by
N-protection to prepare an optically active aziridinecarboxylic acid
derivative
whose imino group is protected with 2-nitrobenzenesulfonyl or

L13 ANSWER 2 OF 3
ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:

1980:639100 HCAPLUS
93:239100
Preparation of fluoro amines by the reaction of aziridines with hydrogen fluoride in pyridine

AUTHOR (S):

of aziridines with hydrogen fluoride in pyridine solution Wade, Tamsir N. Lab. Chim. Struct. Org., Univ. Nice, Nice, 06034, Fr. Journal of Organic Chemistry (1980), 45(26), 5328-33 CODEN: JOCEAH: ISSN: 0022-3263 Journal CORPORATE SOURCE:

DOCUMENT TYPE:

OTHER SOURCE (S):

UAGE: English
R SOURCE(S): CASREAGT 93:239100
HF combines regiospecifically with aziridines to give 2-fluoro
amines in good yields. F attack is in all cases completely
directed to the most substituted ring carbon or to the benzylic carbon.
The results are consistent with an SNI-type mechanism which involves
isomerization of the pos. charged intermediate.

L13 ANSWER 1 OF 3 HCAPLUS COPYRIGHT 2005 ACS on STN (Continued)
4-nitrobenzenesulfonyl and treating this deriv. with an organometallic
reagent or by subjecting an optically active 3-haloalanine deriv. to
N-protection to obtain an optically active 3-haloalanine deriv.
XCH2C*H(NHP1)CO2R2 [X, asterisk = as given above; R2 is hydrogen or the
like; P1 is 2-nitrobenzenesulfonyl or 4-nitrobenzenesulfonyl whose aming
group is protected with 2-nitrobenzenesulfonyl or 4-nitrobenzenesulfonyl
and treating this deriv. with an organometallic reagent. According to
such processes, natural and nonnatural optically active amino acids can

prepd. from inexpensive raw materials through simple and easy operation.

REFERENCE COUNT: 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR

FORMAT

RECORD. ALL CITATIONS AVAILABLE IN THE RE

L13 ANSWER 3 OF 3 HCAPLUS COPYRIGHT 2005 ACS ON STN ACCESSION NUMBER: 1960:118138 HCAPLUS ORIGINAL REFERENCE NO: 54:22552d-1,22553a-1,22554a TITLE: Formation, ring cleavage, and 54:2252d-i,22553a-i,22554a Formation, ring cleavage, and isomerization of ethylenimine-2-carboxylic acid derivatives Gunderman, Karl Dietrich; Holtzmann, Gerhard; Rose, Hans Joachim; Schulze, Helmut Univ. Munster, Germany Chemische Berichte (1960), 93, 1632-43 CODEN: CHBEAM; ISSN: 0009-2940 AUTHOR (S): CORPORATE SOURCE: SOURCE: DOCUMENT TYPE: Journal MENT TYPE: Journal UAGE: Unavailable R SOURCE(S): CASREACT 54:118138
The Kinetic investigation of HCl elimination from H2NCH2CHClCO2H (I) and ClCH2CH(NM2)CO2H (II) in the presence of NaOH showed that 2-carboxyethylenimide (III) was formed in both cases. The hydrolysis of OTHER SOURCE(S): and II at pH 6 yielded in both cases mixts. of DL-serine (IV) and isoserine (V) of identical composition The tendency for the formation of \$\beta\$-substituted-a-amino acids from III increased with increasing nucleophilicity of the cleaving reagent. 1-Benzyl-2-cyanoethylenimine Va rearranged thermally to \$\alpha\$-(N-benzylimino)propionitrile (VI) and PhCH:NCH(CN)Me (VII). CH2:CCICC2Me (76 g.), 8B g. phthalimide, and 200 cc. CGH5 treated with 2.4 g. Na in 176 cc. absolute MeOH, the mixture heated 0.5 h., cooled, filtered after a few hrs., the residue washed with a little MeOH, the combined filtrates evaporated in vacuo, the residue dissolved CHC13, washed with 0.2N NaOH and H2O, evaporated, and the combined htes recrystd. from MeOH yielded 120-3 g. Me -chloro-β-phthalimidopropionate (VIII), m. 125°. VIII and the 10-fold amount of 20% HCl refluxed 5 h., cooled, filtered, evaporated in vacuo, the dissolved in H2O, treated with C, evaporated, dissolved in iso-PrOH, the dissolved in HZO, treated with C, evaporated, dissolved in 1so-PrOH, solution concentrated to beginning crystallization, and diluted with dry Et2O gave 800 I.HCl. Et ester (5 g.) of I.HCl, 10 g. N(CH2CH2OH)3, and 50 cc. absolute EtOH heated 5 ad 5 h. at $60-70^{\circ}$ with stirring, filtered, and distilled in vacuo into a cooled (-80°) receiver gave about 50° Et ester (IX) of III as an alc. solution; the solution treated with 100 cc. MeOH (saturated at 0° NH3), kept overnight, and evaporated yielded 44-8% amide of III, m. 124° (EtOAc-petr. ether). Alc. IX treated with 1 mol equivalent N LiOH, refrigerated 24 h., evaporated in vacuo, the residue evaporated with dry CGH6, dissolved in 50 cc. warm absolute EtOH, the solution cooled, and diluted with

2 vols. dry Et2O gave 0.8-1.0 g. powdery Li salt of III, m. 260-70°
(decomposition), which refluxed in Et0H yielded partially polymeric
materials,
Rf 0.73 (65:35 C6H5N-H2O). Li salt of III in the min. amount of H2O
treated
with the calculated amount of aqueous AgNO3 and diluted with Et0H gave

the Ag salt of
III, pale yellow, which turned gradually brown, even in the dark. Li

Page 9

12/27/2005

- L13 ANSWER 3 OF 3 HCAPLUS COPYRIGHT 2005 ACS on STN (Continued) (0.9 g.) of III in 20 cc. H2O treated dropwise at room temp. with
- stirring
 with 50 cc. 20% H2SO4, the mixt. kept overnight, refluxed 1 h., treated
 with BaCl2, filtered, and purified with Lewatit S-100 gave 0.8 g. crude
 - contg. less than 8% V. A series of similar hydrolyzes was performed with I (pH, temp., % yield of IV + V, % content of IV in product given): 6, 100°, about 80, 83; 6, 60°, about 80, 86-8; 5, 100°, about 80, 80; 2, 100, about 80°, 53. II gave similarly at pH 6 and 100° 75% mixt. of IV and V contg. 83% IV. Li salt (1.1 g.) of III in 20 cc. H2O added dropwise with stirring and cooling to 4 g. ACSH in 40 cc. H2O, the mixt. refluxed 0.5 h., treated with excess 20% HCl, refluxed 8 h., evapd. in vacuo, and the residue treated in 11q. NH3 with Na and PhCH2Cl gave 0.75 g. S-benzyl-Di-isocysteine (X), m. 190-5°. Alc. IX (from 5 g. I.HCl) treated with 2 g. ACSH in 30 cc. abs. EtOH gave X.
- PRCH2C1 gave 0.75 g. S-benzyl-DL-isocysteine (X), m. 190-5°. Alc. IX (from 5 g. I.HC1) treated with 2 g. AcSH in 30 cc. abs. EtoH gave X. portion of the crude residue (after evapn. of the HC1) from a similar run treated with Raney Ni gave B-alanine (XI) contg. very little dealanine; another portion (0.63 g.) of the residue treated with Na and PhcHZC1 in 1ig. NH3 gave about 0.20 g. X contg. traces of S-benzyl-DL-cysteine. I.HC1 (1.6 g.) in 200 cc. H20 neutralized with NaOH, heated to reflux, adjusted with NaOH to pH 7-7.5, concd. to 50 cc., added dropwise with stirring at 20° to 150 cc. N HC1, kept 15 h. at room temp., evapd. in vacuo, the residue extd. with abs. EtoH, and the ext. evapd. gave 83% mixt. of I and II contg. 31% I; a similar run at -4° yielded 80% mixt. contg. 38% I. A run with 90 mol equivs. 6N HC1 at 20° gave 78% mixt. contg. 31% II. I.HC1 converted similarly to III and then cleaved at 20° with N HBr yielded 70% mixt. of HNCH2CHBrCO2H and BrCH2CH(NN2)CO2H (XII) contg. 51-4% XII. A run with N HC1 at -4° yielded 90% mixt. of I and II contg. 39 II. IX treated at -4° with HC1 in Me2CO-Et2O gave 75% mixt. of I and II, contg. 58% I. Na sait of III treated with NHC1, the mixt. neutralized with NHGN, and the product fractionally crystd. from aq. EtOH gave 10% II, decompd. at 142°. Et ester of I.HC1 (5 g.) in 50 cc. Et2O treated with solid K2CO3 to a crystal slush, the aq. phase extd. with Et2O, the combined Et2O solns. dried several hrs. at -4° and then evapd., the crude residual Et ester of I (2.8-3.0 g.) dissolved immediately in abs. EtOH, dild. to 100 cc., and aliquots titrated for chloride ion gave the rate data which was presented graphically. The rates of the HC1 elimination from I and II were detd. similarly and found to be 7.75 + 103 min.-1 and 2.57 + 103 min.-1 at 35.50°, resp.

 **BCH2CHBCRCN (XIII) and the appropriate mine emise equimolar amts.)

 in C6H6 refluxed 3 h. with 2 mol equivs. Et3N gave a mixt. of products. XIII (50 g.), 25 g. PhCH2M2L, 47 g. Et3N, and 400 cc. C6H6 gave thus 24-9 g. m
- L13 ANSWER 3 OF 3 HCAPLUS COPYRIGHT 2005 ACS on STN (Continued)
 XI (N-phthalyl deriv. m. 160°); the alc. ext. concd. gave a small
 amt. of XI; the mother liquors evapd. and the residue dissolved in EtOH
 and dild. with dry Et2O yielded about 10 g. N-PhCH2 deriv. of IV. A
- mixt.

 (5 g.) of Va, VII, and XIIIa heated to 100° and then hydrolyzed with 320 cc. N HCl yielded 2.0 g. BzH (semicarbazone m. 220°) and 1.42 g. XI. Va-VII-XIIIa mixt. (9.4 g.) in 110 cc. 208 HCl kept several hre, at room temp., washed with Et20, evapd., and the residue treated
- with

 120 left 2-3 g. PhCH2NH2.HCl, m. 255* (3,5-dinitrobenzoate m. 210*); the filtrate furnished up to 28% XI. Va-VII-XIIIa mixt. (2.0 g.) in 10 cc. dry Me2CO treated with cooling with 16 cc. about 2N HCL-Et2O, the mixt. refrigerated overnight, dild. with 100 cc. dry Et2O, and filtered after several hrs. gave 2.5-2.7 g. ClCH2CH(CN)NHCH2Ph.HCl
- and filtered after several hrs. gave 2.5-2.7 g. CICR2CH(CNNRKRFFH.RCI and filtered after several hrs. gave 2.5-2.7 g. CICR2CH(CNNRKFFH.RCI is and filtered with KI and alc. RCI in HCDNNe2 liberated iodine. Va-VII-XIIIa mixt. (5.6 g.) and 3.0 g. KOH in 30 cc. ECOH heated 2 h. at 50-60°, the mixt. concd. in vacuo to about 15 cc., dild. with 30 cc. H2O. the aq. phase extd. with Et2O, and the combined org. layer and extd. worked up gave 3.55 g. N-benzylethylenimde-2-carboxamide, m. 112° (ECOHET2O). XIII (50 g.), 32 g. p-MeOCGHANHZ, 47 g. Et3N, and 400 cc. C6H6 yielded in the usual manner 15.2 g. N-(p-methoxyphenyl)-2-cyanoethylenimine (XIV), b0.05 126-7°, n2DD 1.5556. XIV (5.4 g.) hydrolyzed with 290 cc. N HCl yielded 1.3 g. XI and 2.1 g. p-MeOCGH4CHO, which gave 2.5 g. semicarbazone, m. 209-11°. Similarly were prepd. the N-Bu analog (XV) of XIV, 774, b12 85-7°, n2DD 1.4430, N-neopentyl analog of XIV, 674, b14 83-4°, n2DD 1.4430, N-(p-ClC6H4CH2) analog of XIV, leaflets, 66%, m. 69° (ligroine). XV (3.75 g.) treated in the usual manner with 2 g. KOH in 30 cc. EtOH
- 3.0 g. N-butylethylenimine-2-carboxamide, m. 61° (C6H6-petr. ether). MeCHBrCHBrCN treated in the usual manner with PhCH2NH2 yielded 35-6% product, C1H12N2, b0.2 112-15°. The IV-V mixts. were analyzed spectrophotometrically with the absorption max. of the Cu complex salts at 620 and 710 mu.

=> s metal

1616377 METAL

818332 METALS

L14 1961287 METAL

(METAL OR METALS)

=> s 19 and 114

L15 4 L9 AND L14

=> d ibib abs 1-4

L15 ANSWER 1 OF 4
ACCESSION NUMBER:
DOCUMENT NUMBER:
1996:167574 HCAPLUS
124:23221
124:23221
Aziidhe compounds, methods of preparation, and reactions thereof, as intermediates for thiamphenicol and analogs
Davis, Franklin A.: Zhou, Ping: Reddy, Gaddampally Venkat
Drewel University, USA
POT Int. Appl., 62 pp.
CODDENT TYPE:
DOCUMENT TYPE:
LANGUAGE:
English
FAMILY ACC. NUM. COUNT:
1
PATENT INFORMATION:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

OTHER SOURCE(S):

CASREACT 124:232231; MARPAT 124:232231

Novel N-sulfinyl-2-carboxy- and N-hydrogen-2-(hydroxymethyl)aziridine compds. I and II and their stereoisomers are provided (wherein R1-R5 = H, hydrocarbyl radicals containing 1-40 C atoms, 0-40 halo atoms, and 0-10 heteroatoms (B, N, O, S, P, Si, Se); both R3 and R4 * H; sulfinyl molety may be racemic or optically enriched). The asym. synthesis of N-sulfinylaziridines is readily accomplished in high disatereomeric

purity and good yield by a Darzens-type reaction of a metal enclate of

L15 ANSWER 2 OF 4 HCAPLUS COPYRIGHT 2005 ACS ON STN ACCESSION NUMBER: 1986:626320 HCAPLUS DOCUMENT NUMBER: 105:226320

105:226320
Aziridine-2-carboxylic acid salts
Sadao, Kitagawa; Takashi, Yokoi; Mitsumasa, Kaitoh
Research Assoc. for Utilization of Light Oil, Japan
Eur. Pat. Appl., 39 pp.
CODEN: EPXXDW INVENTOR (S):

PATENT ASSIGNEE(S): SOURCE:

DOCUMENT TYPE:

English

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 191462	Al	19860820	EP 1986-101728	19860212
R: DE, FR, GB, JP 61186361	IT A2	19860820	JP 1985-25775	19850213
JP 62019567	A2	19870128	JP 1985-159498	19850719
US 4935527	A	19900619	US 1988-289440	19881223
PRIORITY APPLN. INFO.:			JP 1985-25775 A	19850213
			JP 1985-159498 A	19850719

US 1986-828549

GΙ

The title compds. I (R1-R4 = H, C1-10 hydrocarbyl, M = NH4, mmtal ion; n = valence of M), useful as neoplasm inhibitors, are prepared by

reaction of a 2,3-dihalopropionic acid or an a-haloacrylic acid derivative with aqueous NH3 in presence of an alkaline earth metal hydroxide. ClcH2CHClcO2Me, aqueous NH3, and Ca(OH)2 were charged into an autoclave at 90° for 5 h to give I (R1-R4 = H; N = Ca; n = 2) (95.3% yield).

L15 ANSWER 1 OF 4 HCAPLUS COPYRIGHT 2005 ACS on STN (Continued) an a-halo ester with an enantiopure sulfinimine. Ring-opening of the aziridines affords a-amino acids and otherwise difficult to prep. syn-B-hydroxy-a-amino acids, both key structural units found in many bioactive materials. The N-sulfinyl radical may be selectively renoved from the novel aziridine compds. by treatment with acid or base. Alternatively, the N-sulfinyl radical may be oxidized to provide the corresponding N-sulfonyl-aziridine, or reduced to form the corresponding 1H-2-(hydroxymethyl)aziridine, either of which may subsequently be ring-opened to provide precursors to bloactive compds. For example, BrCH2CO2Me was lithiated with (Me3Si)2NLi in THF, and reacted

For example, Structure was Alliana and Theorem 1. For example, Structure was Alliana and Theorem 2. For example, Structure was a similarly preparation of the meaning of III with R3 = p-(MeS)C6H4 was similarly preparation them reduced to the corresponding hydroxymethyl compd. II, hydrolyzed to an aminopropanediol, N-dichloroacetylated, and oxidized with m-clC6H4C(O)OOH, to give the antibiotic thiamphenicol.

L15 ANSWER 3 OF 4 HCAPLUS COPYRIGHT 2005 ACS ON STN ACCESSION NUMBER: 1986:573039 HCAPLUS DOCUMENT NUMBER: 105:173039 HCAPLUS 1

p-Chioroalanine Kitagawa, Sadao; Yokoi, Takashi; Minafuji, Mitsumasa Keishitau Ruibun Shinyoto Kaihatsu Gijutsu Kenkyu Kumiai, Japan Jpn. Kokai Tokkyo Koho, 4 pp. CODEN: JKXXAF PATENT ASSIGNEE (S):

SOURCE:

Patent

DOCUMENT TYPE: LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. APPLICATION NO. KIND DATE DATE JP 60252453 PRIORITY APPLN. INFO.: A2 19851213

OTHER SOURCE (S): CASREACT 105:173039

B1 19860212

The title compound (I), useful as an intermediate cysteine, was prepared

heating an aziridine derivative I (R = CO2Rl, cyano, CONH2; Rl = H, Cl-5 alkyl, alkali or alkaline earth metal, NH4) with a Cl-containing inorg. salt in an aqueous solvent at pH 0.01-6.0. Thus, heating II (R = CO2Na)

water in the presence of p-Mec6H4SO3H and NaCl at 100° for 3 h gave 89.5% I in 100% conversion.

L15 ANSWER 4 OF 4
ACCESSION NUMBER:
DOCUMENT NUMBER:
PATENT ASSIGNEE(S):
DOCUMENT TYPE:
LANGUAGE:
PAMILY ACC. NUM. COUNT:
PATENT INFORMATION:

LANGUAGE:
PATENT INFORMATION:

LANGUAGE:
PAMILY ACC. NUM. COUNT:
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 57146751	A2	19820910	JP 1981-32501	19810309
JP 60039357	B4	19850905		
PRIORITY APPLN. INFO.:			JP 1981-32501	19810309

Aziridine-2-carboxylic acid (I) salts were prepared by treating β -haloslanines, their esters, or mineral acid salts with alkali (or alkaline earth) metal hydroxides or aqueous NH3 in aqueous media. Thus,

g NaOH in H2O was added to 24 g β -chloroalanine-HCl in H2O at room temperature to give, after 24 h, 92.6% I Na salt. Similarly prepared were I K salt and I Ca salt.

Manine

=> s 16/prep

364 L6

3402624 PREP/RL

L16

209 L6/PREP

(L6 (L) PREP/RL)

=> s 13/rct

2444 L3

2802094 RCT/RL

L17

397 L3/RCT

(L3 (L) RCT/RL)

=> s 116 and 117

L18 6 L16 AND L17

=> d ibib abs 1-6

L18 ANSWER 1 OF 6
ACCESSION NUMBER:
DOCUMENT NUMBER:
135:195786
Processes for preparing optically active amino acid derivatives
Sugawara, Masanobu; Fujii, Akio; Okuro, Kazumi; Saka, Yasuhiro; Nagsahima, Nobuo; Inoue, Kenji; Takeda, Toshihiro; Kinoshita, Koichi; Moroshima, Tadashi; Fuse, Yoshihide; Ueda, Yasuyoshi
Kaneka Corporation, Japan; et al.
CODE:
DOCUMENT TYPE:
LANGUAGE:
FAMILU ACC. NUM. COUNT:
PATENT INFORMATION:
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LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

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110	5720	449	14		82		2003	0413		03 2	002-	3203				20020	,203	
119	6720 2005	1435	96		A 1		2005	0530		115 2	003-	7164	30			20031	120	
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										WO 2	001-	JP11	32	,	w	20010	216	
										us 2	002-	9263	46		A.3	20020	205	

OTHER SOURCE(s): CASREACT 135:195786; MARPAT 135:195786

AB An optically active amino acid derivative is prepared either by subjecting an optically active 3-haloalanine derivative XCH2C*H(NH2)CO2R1 [X is halogen; Rl is hydrogen or the like; the asterisk represents an asym. carbon atom] to N-protection followed by cyclization or cyclization followed by N-protection to prepare an optically active aziridinecarboxylic acid derivative whose imino group is protected with 2-nitrobenzenesulfonyl or

L18 ANSWER 2 OF 6 HCAPLUS COPYRIGHT 2005 ACS ON STN ACCESSION NUMBER: 1994:580151 HCAPLUS
DOCUMENT NUMBER: 121:180151

The synthesis and stability of aziridino-glutamate, TITLE:

irreversible inhibitor of glutamate racemase Tanner, Martin E.; Miao, Shichang Dep. Chem., Univ. British Columbia, Vancouver, BC, AUTHOR (S): CORPORATE SOURCE:

SOURCE:

121, Can. Tetrahedron Letters (1994), 35(24), 4073-6 CODEN: TELEAY; ISSN: 0040-4039 Journal English CASREACT 121:180151

DOCUMENT TYPE:

OTHER SOURCE(S):

Aziridino-glutamate (\pm)-I was synthesized by heating α -fluoromethylglutamate II in base. In neutral solution, (\pm)-I was shown to cyclize to the γ -lactone III with a half life of 4 min. Aziridino-glutamate was shown to irreversibly inactive glutamate racemase by alkylating an active site cysteine residue. Electrospray mass spectrometry was used to establish that a covalent bond had formed and that this bond protects one of the enzyme's two cysteine residues from reacting with iodoacetate under denaturing conditions.

L18 ANSWER 1 OF 6 HCAPLUS COPYRIGHT 2005 ACS on STN (Continued)
4-nitrobenzenesulfonyl and treating this deriv. with an organometallic reagent or by subjecting an optically active 3-haloslanine deriv. to N-protection to obtain an optically active 3-haloslanine deriv. to CRCC*H(NNP1)COZR2 [X, asterisk = as given above: R2 is hydrogen or the like: P1 is 2-nitrobenzenesulfonyl or 4-nitrobenzenesulfonyl] whose amino group is protected with 2-nitrobenzenesulfonyl or 4-nitrobenzenesulfonyl and treating this deriv. with an organometalic reagent. According to such processes, natural and nonnatural optically active amino acids can be

prepd. from inexpensive raw materials through simple and easy operation.

REFERENCE COUNT: 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE

L18 ANSWER 3 OF 6 HCAPLUS COPYRIGHT 2005 ACS ON STN ACCESSION NUMBER: 1990:459808 HCAPLUS DOCUMENT NUMBER: 113:59808

2-(4-Amino-4-carboxybutyl)aziridine-2-carboxylic TITLE: acid.

A notent irreversible inhibitor of diaminopimelic acid

AUTHOR(S):

CORPORATE SOURCE:

epimeráse. Spontaneous formation from α-(halomethyl)diaminopimelic acids Gerhart, Fritz; Higgins, William; Tardif, Chantal; Ducep, Jean Bernard Strasbourg Cent., Merrell Dow Res. Inst., Strasbourg, 67084, Fr. Journal of Medicinal Chemistry (1990), 33(8), 2157-62 CODEN: JMCMAR; ISSN: 0022-2623 Journal SOURCE:

DOCUMENT TYPE: Journal

LANGUAGE: OTHER SOURCE(S): English CASREACT 113:59808

The title compound (I) was identified as the product of spontaneous hydrolysis of α -(halomethyl)diaminopimelic acids RCH2C(NN2) (CO2H)CH2CH2CH(NN2) CO2H) CH2CH2CH(NN2) CO2H (II, R = F, Cl, Br). Under physiol. conditions, I is an extremely potent irreversible inhibitor of the bacterial enzyme diaminopimelic acid epimerase (EC 5.1.1.7). This

mode of action of an α -halomethyl amino acid with a nonpyridoxal enzyme is investigated. Synthesis and characterization of I and II, kinetics of spontaneous formation of I from II, and kinetics of enzyme inhibition by both I and II are reported.

L18 ANSWER 4 OF 6 HCAPLUS COPYRIGHT 2005 ACS on STN
ACCESSION NUMBER: 1986:626320 HCAPLUS
DOCUMENT NUMBER: 105:226320
Aziridine-2-carboxylic acid salts
INVENTOR(5): Sadao, Kitagawa: Takashi, Yokoi; Mitsumasa, Kaitoh
Research Assoc. for Utilization of Light Oil, Japan
EUR. Pat. Appl., 39 pp.
COOMENT TYPE: Patent
LANGINGF: PRINTED

DOCUMENT TYPE: LANGUAGE:

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 191462	A1	19860820	EP 1986-101728	19860212
R: DE, FR, GB,	IT			
JP 61186361	A2	19860820	JP 1985-25775	19850213
JP 62019567	A2	19870128	JP 1985-159498	19850719
US 4935527	A	19900619	US 1988-289440	19881223
PRIORITY APPLN. INFO.:			JP 1985-25775 A	19850213
			JP 1985-159498 A	19850719
			119 1986-928540 91	10060212

GI

The title compds. I (R1-R4 = H, C1-10 hydrocarbyl, M = NH4, metal ion; n

valence of M), useful as neoplasm inhibitors, are prepared by the

valence of M), useful as neoplasm inhibitors, are prepared by the reaction of a 2,3-dihalopropionic acid or an α -haloacrylic acid derivative with aqueous NN3 in presence of an alkaline earth metal hydroxide. CICHZCHCICOZMe, aqueous NN3, and Ca(GN122 were charged into an autoclave at 90° for 5 h to give I (R1-R4 = H; M = Ca; n = 2) (95.3% yield).

L18 ANSWER 6 OF 6
ACCESSION NUMBER:
DOCUMENT NUMBER:
1978:436591 HCAPLUS
99:36591
99:36591
35-Carboxamido-4-amino-3-isoxazolidone, an asparagine
analog
AUTHOR(S):
CORPORATE SOURCE:
SOURCE:
DOCUMENT TYPE:
LANGUAGE:
CI

HCAPLUS COPPRIGHT 2005 ACS on STN
1978:436591 HCAPLUS
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1978:436591 HCAPLUS
SCIENCE OF STN

DOCUMENT TYPE: LANGUAGE: GI

The synthesis of the title compound I [66620-06-2] from trans-diethyl aziridine-2,3-dicarboxylate [66619-94-1] via diethyl β -chloroaspartate hydrochloride [66619-95-2] is described. Neither I nor the aziridine hydroxamate intermediate II [66620-05-1] had antitumor activity in mice against L1210 and P-388 tumors. I was inactive as an inhibitor of asparagine synthetase from Novikoff hepatoma and did not inhibit the growth of bacteria or fungi.

L18 ANSWER 3 OF 6
ACCESSION NUMBER:
DOCUMENT NUMBER:
PATENT ASSIGNEE(S):
DOCUMENT TYPE:
LANGUAGE:
PATENT ACC. NUM. COUNT:
PARTENT INFORMATION:

DOCUMENT TYPE:
LANGUAGE:
PATENT INFORMATION:

1983:125854
HCAPLUS
38:125854
HCAPLUS
38:125854
HCAPLUS
38:125854
HCAPLUS
Mitaul Toatsu Chemicals, Inc., Japan
Jpn. Kokai Tokkyo Koho, 4 pp.
CODEN: JKCKVAF
Patent
Japanese
1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 57146751	A2	19820910	JP 1981-32501	19810309
JP 60039357	B4	19850905		
PRIORITY APPLN. INFO.:			JP 1981-32501	19810309

AB Aziridine-2-carboxylic acid (I) salts were prepared by treating β-haloalanines, their esters, or mineral acid salts with alkali (or alkaline earth) metal hydroxides or aqueous NH3 in aqueous media. Thus, 20 g NaOH in

H20 was added to 24 g β-chloroalanine-HCl in H2O at room temperature to give, after 24 h, 92.6% I Na salt. Similarly prepared were I K salt and I

Ca salt.

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---Logging off of STN---

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Executing the logoff script...

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COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	66.70	389.57
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	SESSION
CA SUBSCRIBER PRICE	-10.95	-10.95

STN INTERNATIONAL LOGOFF AT 12:36:29 ON 27 DEC 2005